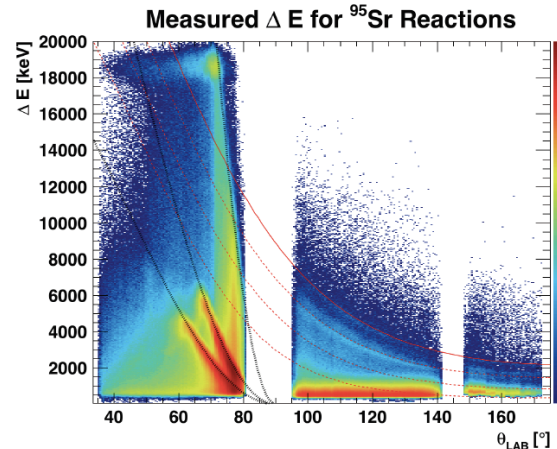
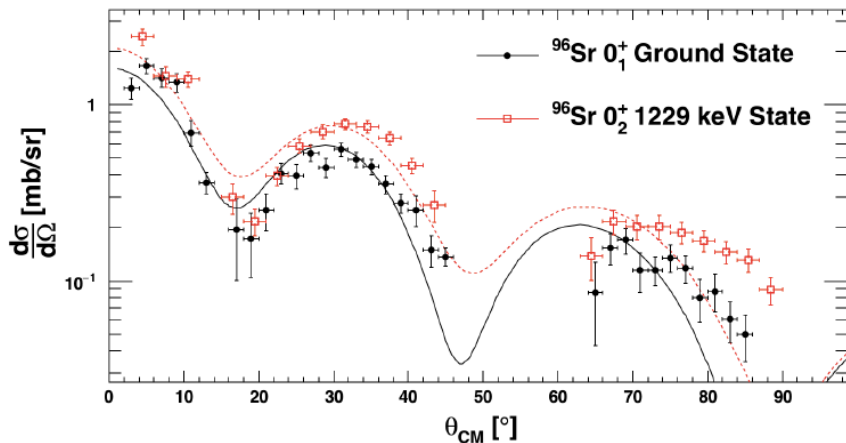
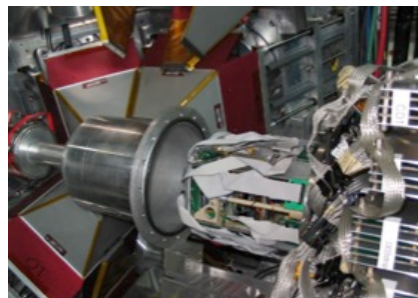


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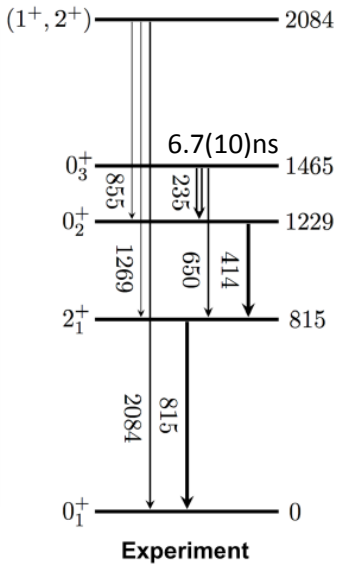
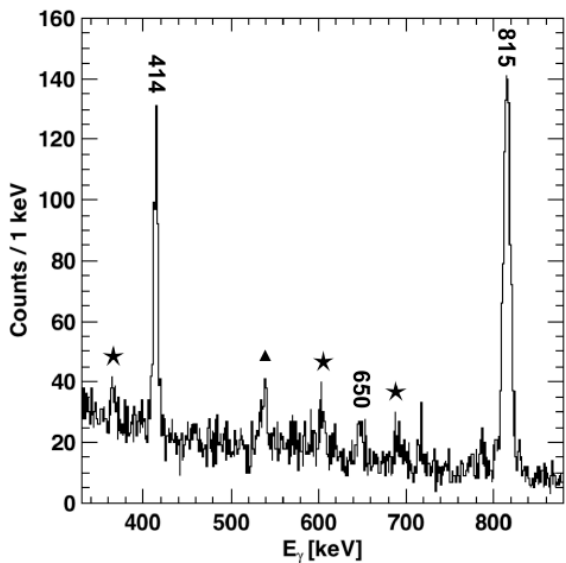
$d(^{95}\text{Sr}, p)^{96}\text{Sr}$ , 5.4 MeV/u,  $\sim 10^7$ pps  
TIGRESS+SHARC

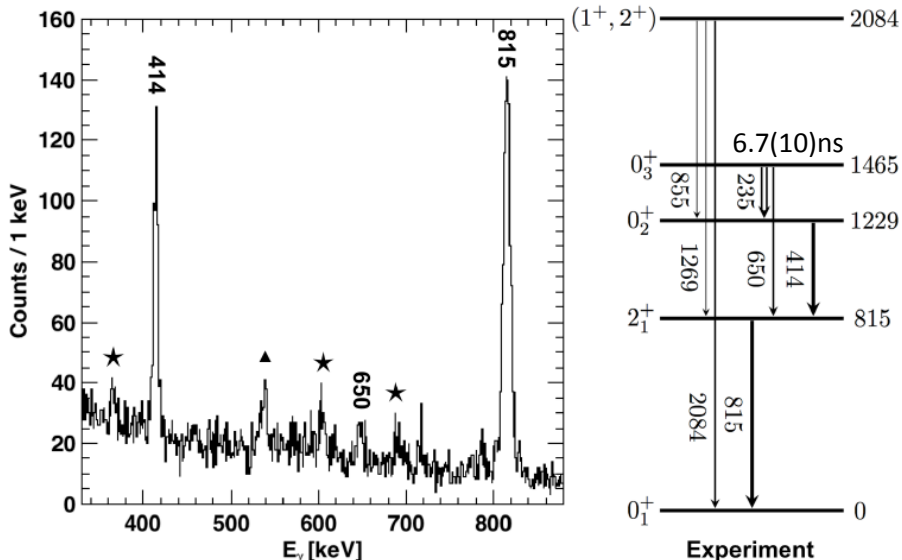


$8^+$	3125
$6^+$	2466
$4^{(+)}$	2120
$4^+$	1975
$4^+$	1793
$2^+$	1507
$0^+$	1229
$2^+$	815
$0^+$	0

$^{96}\text{Sr}_{58}$

185





$$\rho^2(E0) = \left(\frac{3}{4\pi}\right)^2 Z^2 a^2 (1 - a^2) [\Delta(\beta^2)]^2$$

$$a^2 = 0.40(14) \text{ and } \beta_{\text{def}} = 0.31(3), V_{\text{mix}} = 113 \text{ keV}$$

Shell model, proton valence space:

glek a: protons inert in  $[1p_{3/2}]^4$

glek b: excitations allowed to  $[1p_{3/2}]^2 [1p_{1/2}]^2$

glek c:  $0g_{9/2}$  allowed (max 2)

Calculations work well for  $C^2S$  in  $d(^{94}\text{Sr}, p)$ .

TABLE I: Comparison of experimental to calculated spectroscopic factors ( $C^2S$ ) for  $0^+$  states in  $^{96}\text{Sr}$  populated via the  $d(^{95}\text{Sr}, p)$  reaction (more details in the text).

Exp.		Unmixed		glek Ⓐ		glek Ⓑ		glek Ⓒ	
$E_x$ [keV]	$C^2S$	$E_x$ [keV]	$C^2S$	$E_x$ [keV]	$C^2S$	$E_x$ [keV]	$C^2S$	$E_x$ [keV]	$C^2S$
0	0.19(3)	0	0.19(3)	0	1.742	0	1.575	0	1.455
1229	0.23(3)	1314	0	-	-	-	-	-	-
1465	0.34(13)	1380	0.56(23)	2271	0.056	1691	0.098	444	0.105

$${}^{96}\text{Sr } 0_3 - 0_2, V_{\text{mix}} = 113\text{keV}$$

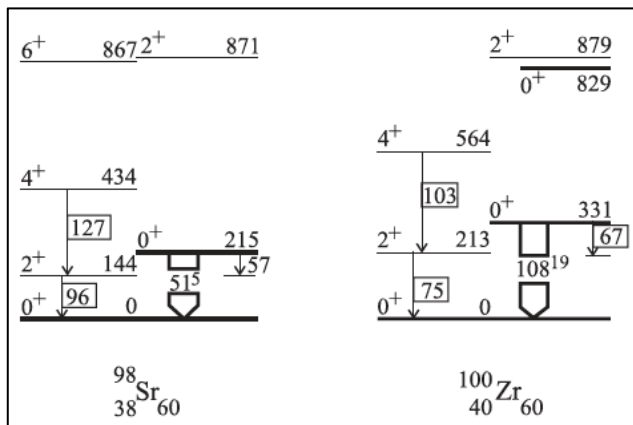
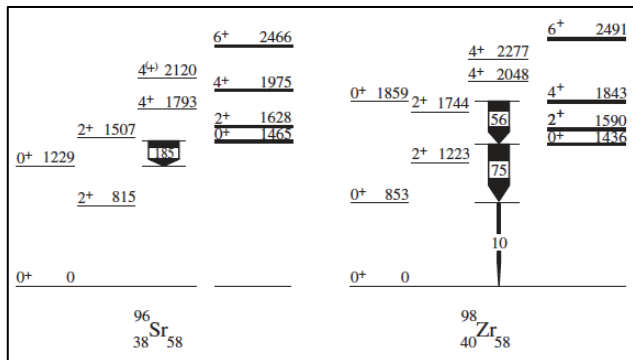


TABLE VI. Mixing strength (in units of keV) used in the description of energy, decay, and transfer reaction properties of coexisting structures.

Isotope	$V_{\text{mix}}$	Quantities fitted	Reference
${}^{72}\text{Kr}$	310	$E$	Becker <i>et al.</i> , 1999; Korten, 2001; Bouchez <i>et al.</i> , 2003
${}^{74}\text{Kr}$	340	$E$	Becker <i>et al.</i> , 1999; Korten, 2001; Bouchez <i>et al.</i> , 2003
${}^{76}\text{Kr}$	250	$E$	Becker <i>et al.</i> , 1999; Korten, 2001; Bouchez <i>et al.</i> , 2003
${}^{78}\text{Kr}$	200	$E$	Becker <i>et al.</i> , 1999; Korten, 2001; Bouchez <i>et al.</i> , 2003
${}^{98}\text{Sr}$	67	$E, B(E2), \rho^2(E0)$	Mach <i>et al.</i> , 1989
	34	$B(E2), \rho^2(E0)$	Wu, Hua, and Cline, 2003
${}^{100}\text{Zr}$	115	$E, B(E2), \rho^2(E0)$	Mach <i>et al.</i> , 1989
	88	$B(E2), \rho^2(E0)$	Wu, Hua, and Cline, 2003
${}^{98}\text{Mo}$	326	$B(M1)$	Rusev <i>et al.</i> , 2005
${}^{100}\text{Mo}$	321	$B(M1)$	Rusev <i>et al.</i> , 2005
${}^{112,114}\text{Cd}$	297	$\sigma(t, p)$	O'Donnell, Kotwal, and Fortune, 1988
${}^{152}\text{Sm}$	310	$\rho^2(E0)$	Kulp <i>et al.</i> , 2007
${}^{176}\text{Pt}$	180	$E$	Dracoulis <i>et al.</i> , 1986
${}^{178}\text{Pt}$	210	$E$	Dracoulis <i>et al.</i> , 1986
${}^{180}\text{Pt}$	220	$E$	Dracoulis <i>et al.</i> , 1986
${}^{182}\text{Pt}$	230	$E$	Dracoulis <i>et al.</i> , 1986
${}^{184}\text{Pt}$	240	$E$	Dracoulis <i>et al.</i> , 1986
${}^{186}\text{Pt}$	220	$E$	Dracoulis <i>et al.</i> , 1986
${}^{188}\text{Pt}$	400	$E$	Dracoulis <i>et al.</i> , 1986
${}^{192}\text{Pb}$	52	$B(E2), \rho^2(E0)$	Van Duppen, Huyse, and Wood, 1990
${}^{194}\text{Pb}$	51	$B(E2), \rho^2(E0)$	Van Duppen, Huyse, and Wood, 1990