

DE LA RECHERCHE À L'INDUSTRIE



INTEGRAL MEASUREMENT OF THE $^{235}\text{M}\text{U}$ ISOMER BY NEUTRON

INELASTIC SCATTERING

G.BÉLIER, V.MÉOT, P. ROMAIN CEA/DAM-DIF, F-91297 ARPAJON, FRANCE

E.BOND, D.J.VIEIRA, J.B.WILHELMY

LOS ALAMOS NATIONAL LABORATORY, LOS ALAMOS, NM 87545, USA

J.A.BECKER, R.MACRI

LAWRENCE LIVERMORE NATIONAL LABORATORY, LIVERMORE, CA 94550, USA

N.AUTHIER, D.HYNECK, Y.JANSEN, J.LEGENDRE, X.JACQUET

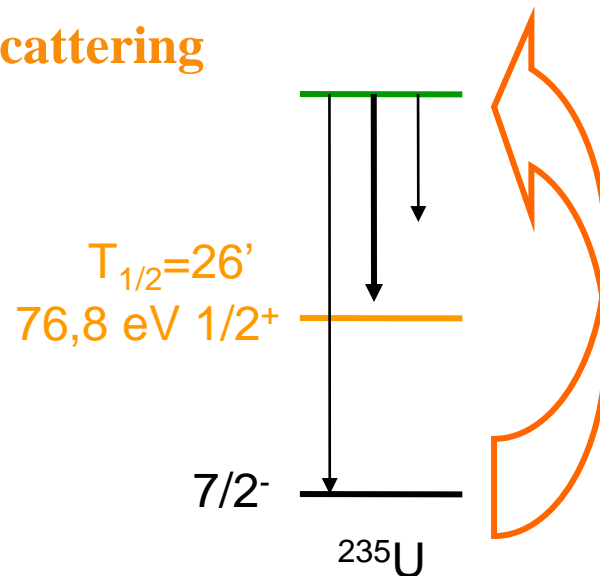
CEA, DAM, VALDUC, F-21120 IS-SUR-TILLE, FRANCE

Fission cross section of the isomer:

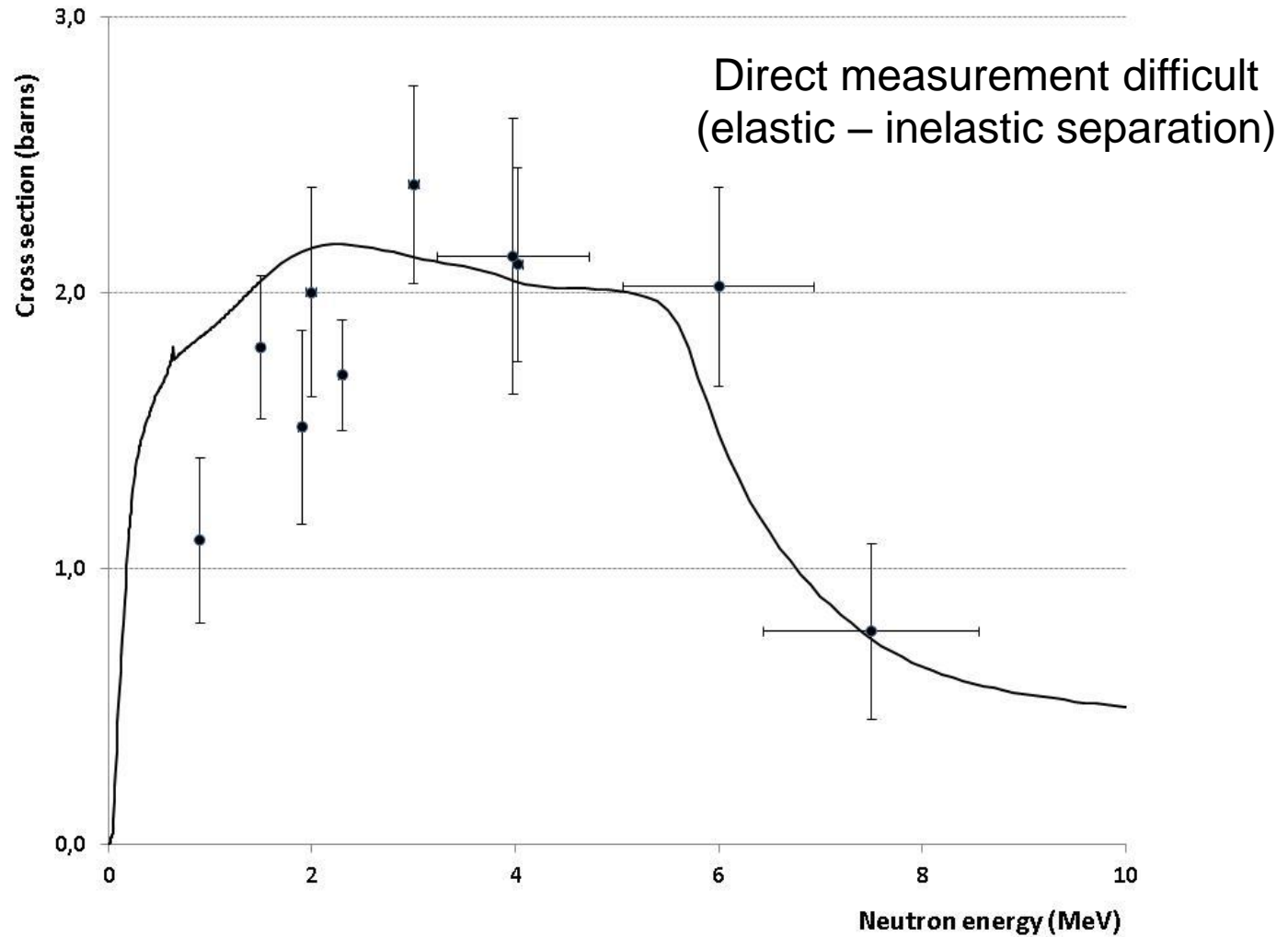
- Experiment: A.D'Eer & al. Phys. Rev. C38(1988)1270 $\rightarrow \sigma^m / \sigma^{gs} = 2,5$ at 25,3 meV
- Theory: J.E.Lynn & al. LA-UR-01-426(2001)21 $\rightarrow \sigma^m < \sigma^{gs}$ $E_n < 0.5$ MeV

Isomer excitation?

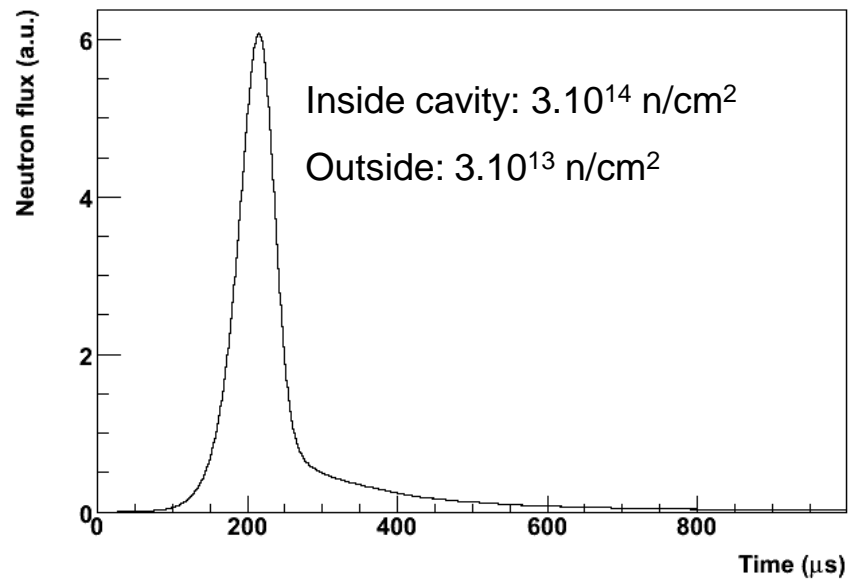
- Electromagnetic processes– V.Méot & al. CEA report R-5944
- **Neutron inelastic scattering**



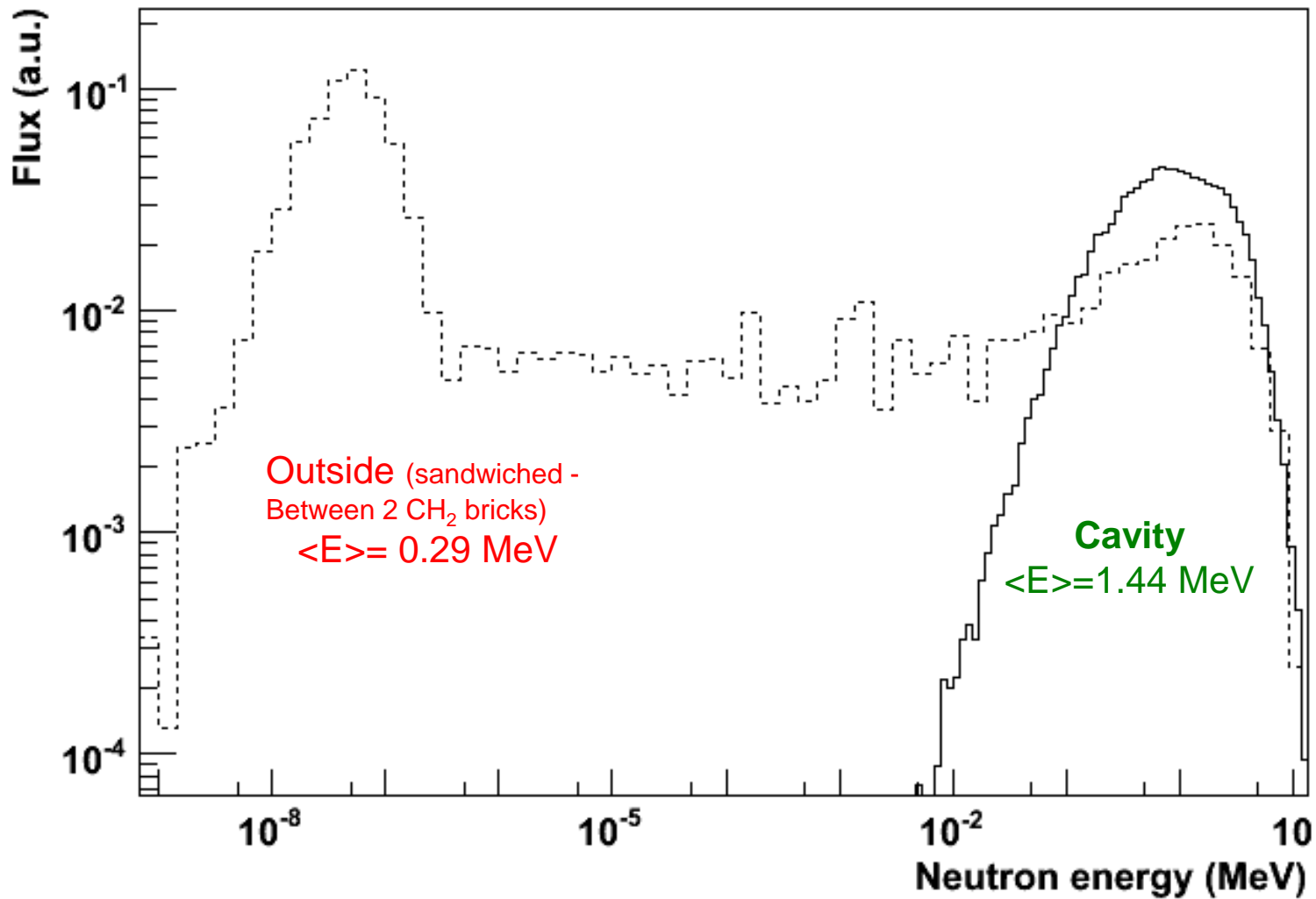
Total inelastic neutron scattering



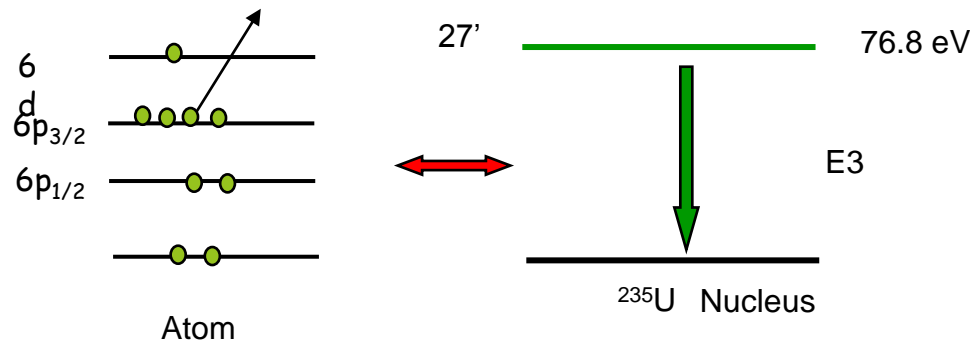
THE CALIBAN CRITICAL ASSEMBLY - VALDUC



- ✓ Shot duration $\sim 60 \mu\text{s}$
- ✓ Fast sample retrieval ~ 30 minutes



Very low energy transition + high multipolarity \rightarrow completely converted



Atomic sub-shell	$6p_{1/2}$	$6p_{3/2}$	6d
Kinetic electron energy (eV)	42.6 ± 0.5	52.6 ± 0.5	64.4 ± 0.5

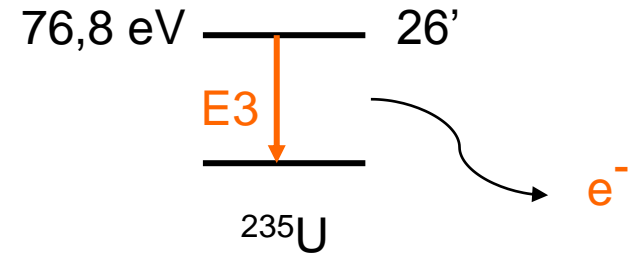
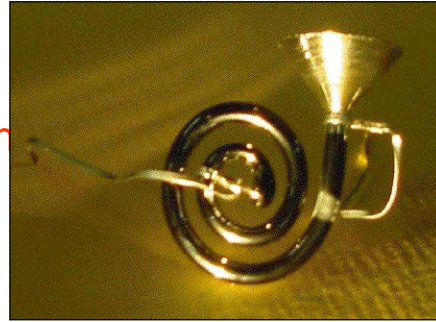
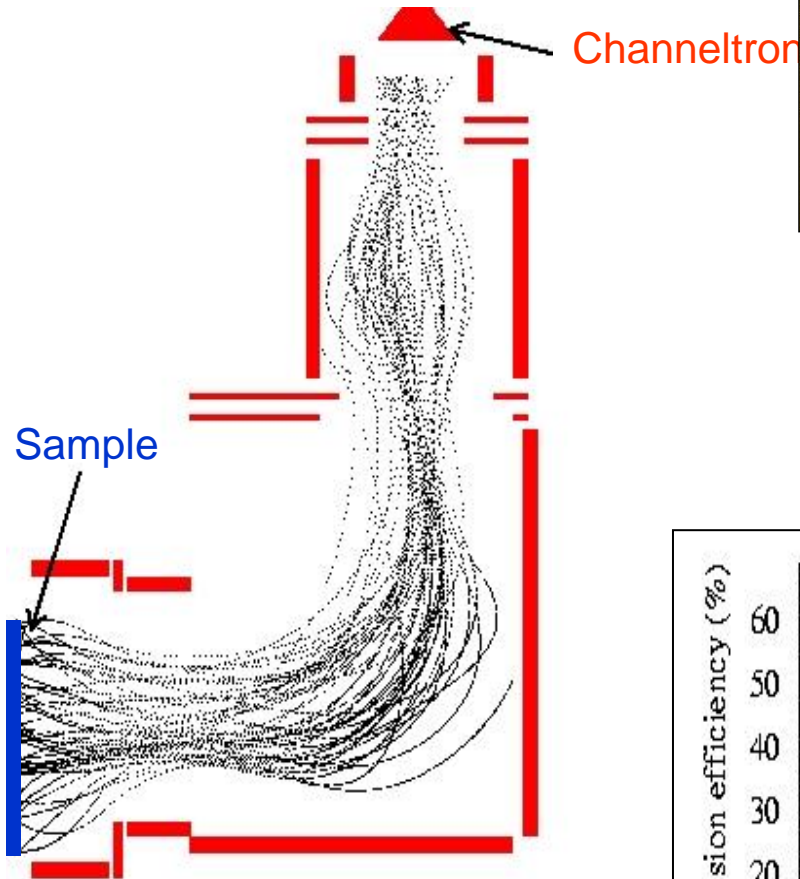
➔ Electrons of very low energies ($E_e \leq 10$ eV) have to be detected

➔ Energy very degraded

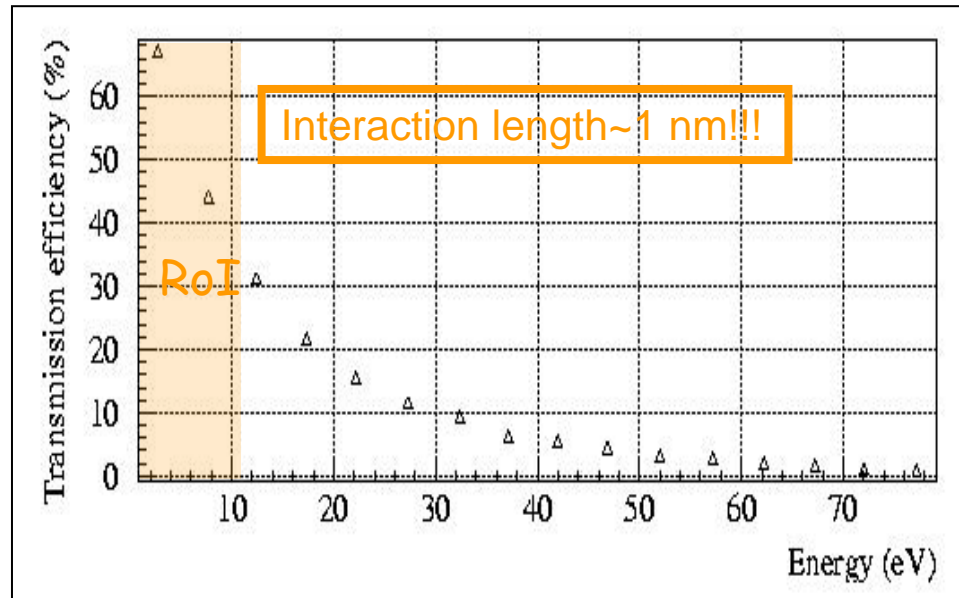
➔ Isomer identification by its period only

Half life depends on the chemical environment !!!! ~ 26 minutes

ELECTRON DETECTOR

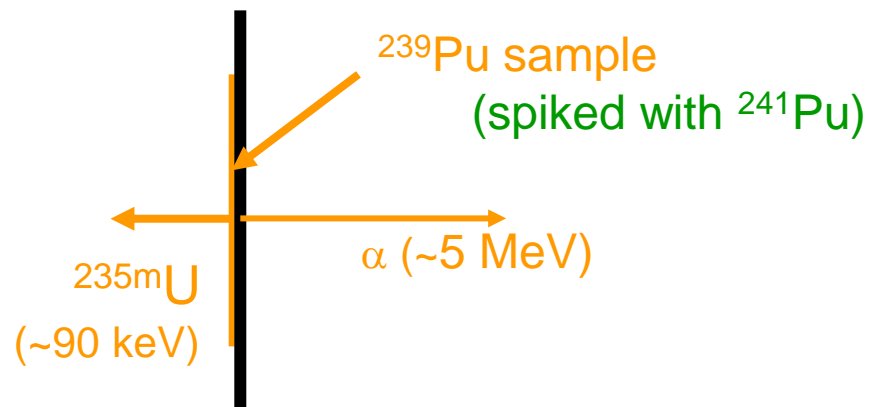


E.Sauvan, V.Méot et J.C. Baudin

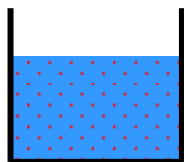


1. Isomer collection

NaCl deposit
1 mg/cm²



2. Electrodeposition



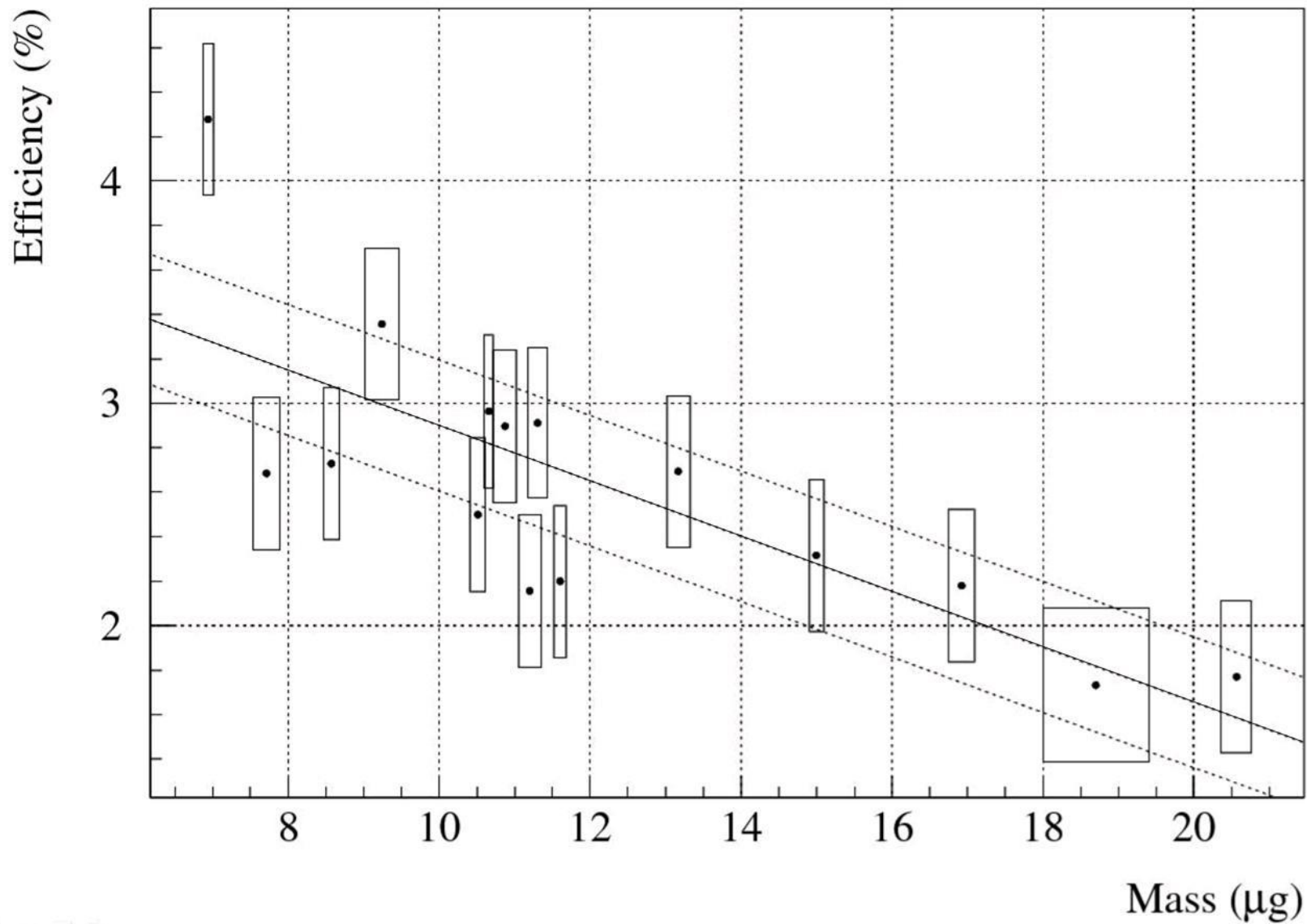
+ ²³⁵U (10 μg)



Deposit ^{235m}U + ²³⁵U
titanium foil



EFFICIENCY CURVE

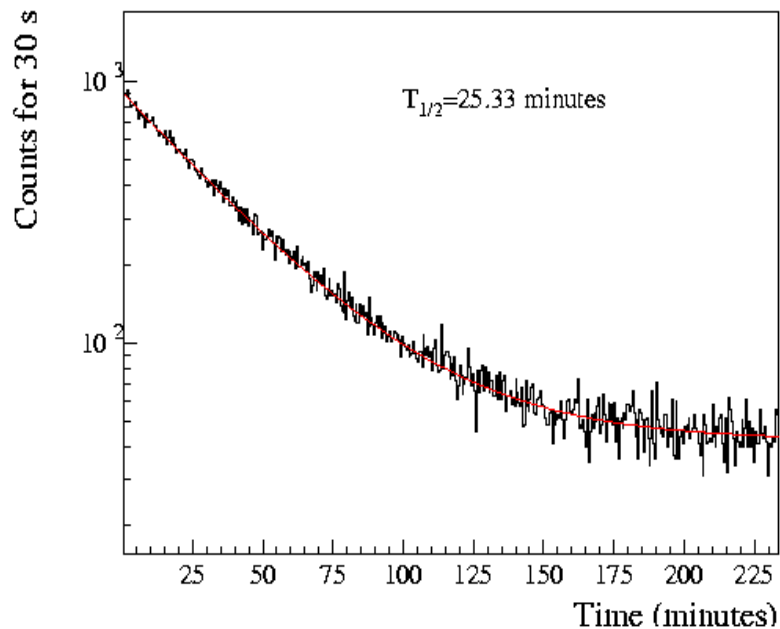


ISOMER HALF-LIFE ?

*M. Neve de Mevergnies and P. Del Marmol, Physics Letters 49B(1974)428
Effect of the oxidation state on half-life of ^{235m}U*

➔ Variation of 10 % depending of the oxidation state

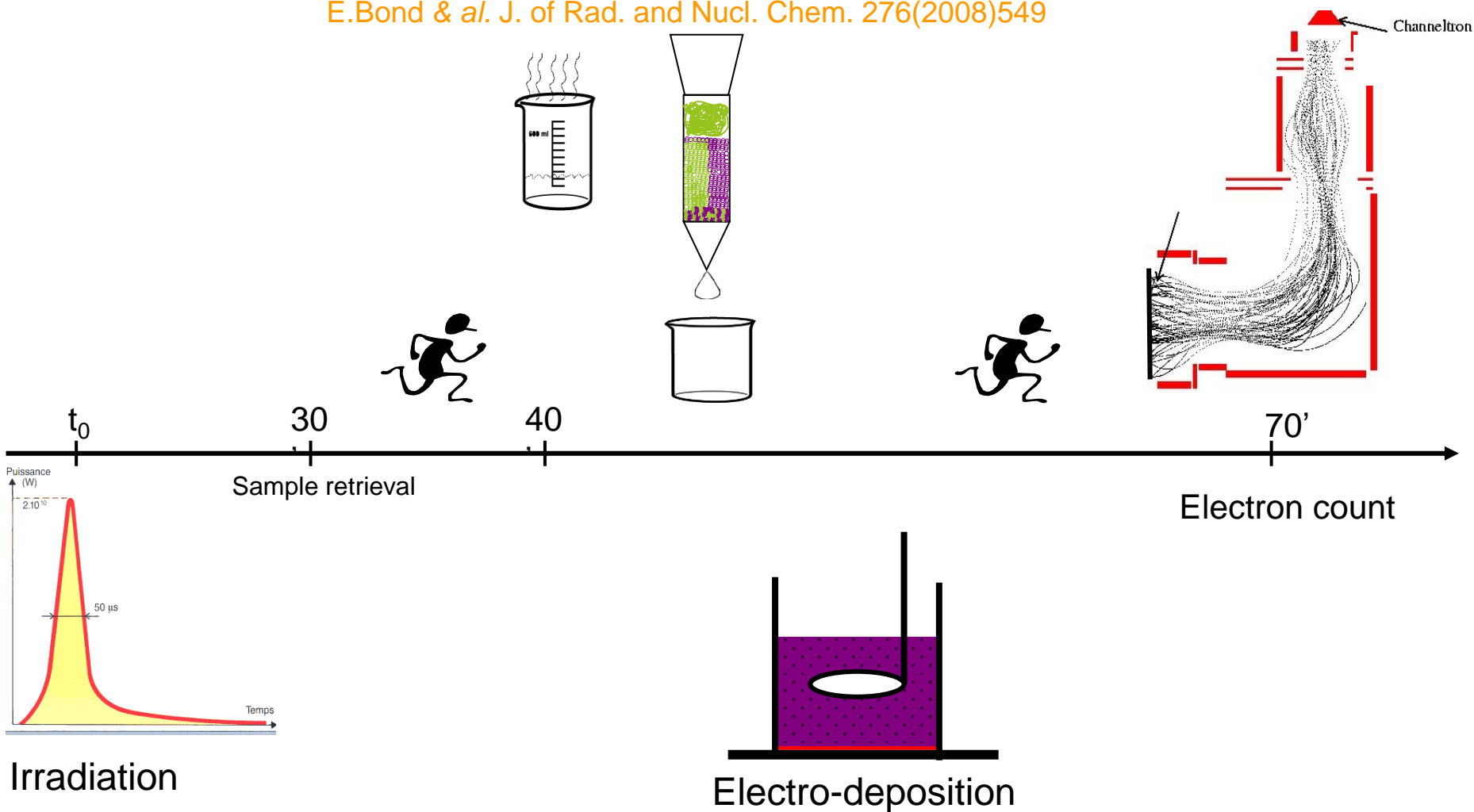
Our measurements:



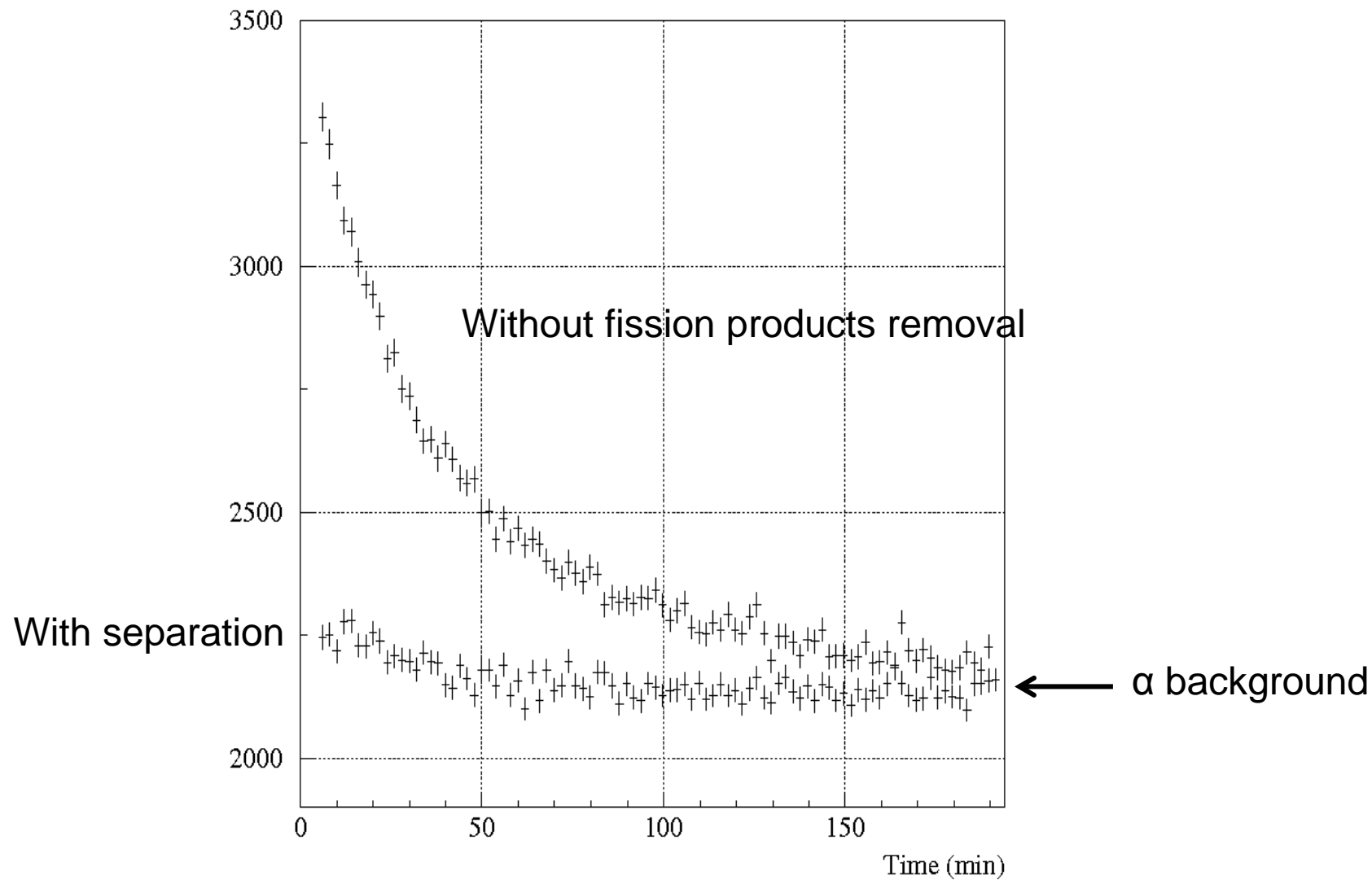
Support	Type of deposit	Half-life (minutes)
Al	Implanted	26,76±0,04
Ti	Implanted	27,4±0,7
Ti	Deposited	25,46±0,04
Pt	Deposited	26,37±0,05
Ag	Implanted	25,7±0,2
NaCl	Implanted	29,01±0,24
Stainless steel	Deposited	25,6±0,1

CALIBAN SHOTS TIMELINE

Chemical separation uranium – produits de fission
E.Bond & al. J. of Rad. and Nucl. Chem. 276(2008)549



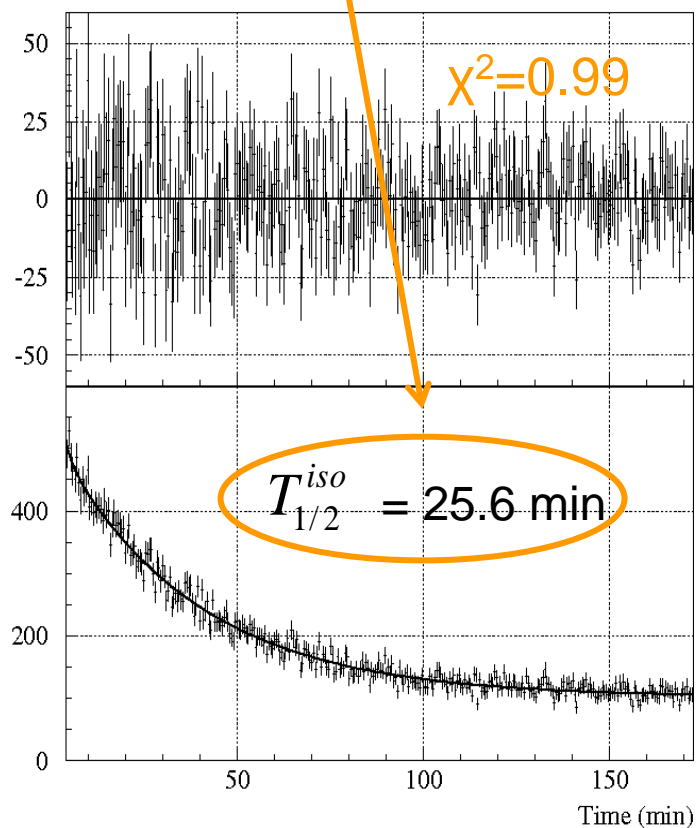
CHEMICAL SEPARATION: TEST WITH ^{236}U



ISOMER ACTIVATION: CAVITY SHOT WITH ON U235

$$f(t) = P_1 + P_2 \exp(-t / T_{1/2}^{iso} \cdot \ln(2)) + P_3 * \exp(-t / T_{1/2}^{bckg} \cdot \ln(2))$$

Half-life fixed to its measured value



1.00 ± 0.13 barns

Parameter	Relative uncertainties (%)
Detection efficiency	7.7
Isomer half-life in aqueous solution	5
Neutron flux measurement	4.7
Samples masses	3.2
Activation measurement	1
Half-lives associated to cooling	0.3

CONCLUSION

